USE LESS CO₂ IN DOUBLE POLY HOUSES

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A carbon dioxide (CO_2) concentration of 1000-1500 ppm is generally suggested in greenhouses to improve productivity. To provide this level in a glass house with one air exchange per hour, 1500 ppm is added per hour.

The air exchange rate in double polyethylene glazed houses may be less than once per hour; it may be only once in several hours if the house is very tight. An **injection rate of only 800-1000 ppm per hour** may be more appropriate. How much is this?

A common size for a plastic covered greenhouse is 2500 sq ft (26x96). If an average height of 8 ft is assumed, the volume is 20,000 cu ft. An injection rate of 1000 ppm means that 20 cu ft of CO_2 would be injected per hour. Since CO_2 weighs roughly 2 oz/cu ft, this is (2x20) 40 oz, or $2\frac{1}{2}$ lbs for a 2500 sq ft greenhouse. (Note that this is 1 lb/1000 sq ft.)

A gallon of propane weighs 4.24 lbs. It provides 12.7 lbs CO_2 . This is enough for 5 hrs at 2.5 lbs/hr, so a CO_2 generator should burn $\frac{1}{5}$ gal/hr for a 2500 sq ft double poly greenhouse.

Combustion of $\frac{1}{5}$ gal propane will produce about 18,000 BTU (91,000 \div 5). This is an advantage on any day when heat is required. This unvented combustion is supposedly 100% efficient and may cost less than providing heat with conventional heating systems using fuel that costs less than propane (since they may be only 60-85% efficient).

A commercially available unit that produces 20,000 BTU/hr at minimum setting is only a smidgin too large for a 2500 sq ft double poly greenhouse. It is acceptable.

A simple gas hot plate produces somewhere around 5000 BTU/hr. Three of these, lit 30-60 minutes before sunrise and turned off an hour before sunset (and during venting) should provide about 850 ppm CO_2/hr . This is probably sufficient for adequate crop response. But some burners are larger or smaller than this. If combustion is not perfect, gaseous by-products may be dangerous. Beware of open flames. On the other hand, as Dr. Sharvelle of Purdue once said, brew your coffee and broil your steaks over the fire: increased profits will pay for them.

steaks over the fire; increased profits will pay for them. Don't worry about the CO_2 distribution. Place the units anywhere that is convenient. Horizontal air flow will give perfect distribution. Even a fan-jet will do a pretty good job. Without any fan, the CO_2 will diffuse the length of the house in less than an hour.

Providing 18,000 BTU/hr to 2500 sq ft will raise temperatures during the day, perhaps as much as 7°. This will supplement normal heating requirements. But it may raise the temperature above the optimum for plant growth. Remember that day temperatures are elevated by 10-15° when CO₂ is used. This warms the crops, soil and everything in the greenhouse, storing energy. This stored heat is returned during the night, reducing heating requirements probably enough to pay for the CO₂.

Suppose that the temperature is being maintained at 60° nights. Normally, ventilation would begin at 70° and full vents used at 75° . With CO₂, day temperatures would be allowed to soar to 80-85° before venting. If fan/tube venting is used with perhaps 6-15 air exchanges per hour, the CO₂ might remain on. If full venting is necessary to keep the temperature low enough, the CO₂ is discontinued.

There is some concern over increased humidity due to water produced as a product of combustion. Don't worry about the extra humidity. Dr. Aldrich (CT Greenhouse Newsletter 108:1-6, Nov. 1981) states that a mature crop will transpire about 0.15 lbs water/hr/sq ft during the day. Assuming 2000 sq ft of plants in a 2500 sq ft greenhouse, this is 300 lbs/hr. The burner, while producing $2\frac{1}{2}$ lbs CO₂ from propane, will produce only 1.4 lbs water. This is inconsequential (for natural gas, the figure is about 2 lbs).

At the same time, remember that humidity is a problem and good air movement is essential to reduce disease problems in all tight greenhouses.

If any of our readers have ideas regarding small CO_2 generators appropriate for greenhouse use, we would be pleased to hear from you.

SOME APPROXIMATE EQUIVALENTS 1000 sq ft x 8 ft deep = 8000 cu ft 8000 cu ft x 1000 ppm == 8 cu ft $8 \text{ cu ft } \text{CO}_2 \text{ x } 2 \text{ oz/cu ft} = 16 \text{ oz} = 1 \text{ lb}$ (more accurately, 8.7 cu ft = 1 lb at 60° F) Therefore, 1000 ppm $CO_2/1000$ sq ft = 1 lb/1000 sq ft One gal propane = 4.24 lbs One pound propane yields 3 lbs CO₂ Therefore, 1 gal = 12.7 lbs CO₂ One gal propane yields 91,000 BTU $91,000 \div 12.7 = 7000 \text{ BTU/lb } \text{CO}_2 \text{ from propane}$ (7 BTU/sq ft)8 cu ft natural gas = 8 cu ft $CO_2 = 16$ oz CO_2 8 cu ft natural gas = 8000 BTU (8 BTU/sq ft) A single poly greenhouse will require about 100 BTU/ sq ft (assuming a 60° difference in indoor/outdoor temperature) A double poly greenhouse will require about 60 BTU/ sq ft (This is 1 BTU/degree temperature difference) Therefore, a 7 BTU/sq ft input should increase the temperature about 7° (8° for natural gas). Laverack & Haines, Inc. 135 Delaware Avenue, Buffalo, N.Y. 14202 **Executive Park East** 890 Seventh North Street Albany, N.Y. 12203 Liverpool, N.Y. 13088

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